



INTEGRATED SCIENCE

STAGE 3

FORMULAE AND DATA SHEET 2012

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This document is valid for teaching and examining until 31 December 2012.

INTEGRATED SCIENCE STAGE 3

Work $W = F_S$; $W = \Delta E$

Potential energy $E_p = mgh$

 $E_k = \frac{1}{2} m v^2$ Kinetic energy

 $P = \frac{W}{t}$ Power

Electrical power $P = VI = I^2R$

 $efficiency = \frac{energy \ out}{energy \ in} \times 100\%$ Efficiency

Percentage composition by mass for ore A_yB_z

% A in ore =
$$\frac{y \times \text{atomic weight of A}}{(y \times \text{atomic weight of A}) + (z \times \text{atomic weight of B})} \times 100$$

% B in ore =
$$\frac{z \times \text{atomic weight of B}}{(y \times \text{atomic weight of A}) + (z \times \text{atomic weight of B})} \times 100$$

Relevant units and definitions

Volumes are given in the units of litres (L), or millilitres (mL). Volume:

2

Energy change: Energy changes are given in the SI unit joule (J).

Population density: Number of an individual species living in a particular place at a particular

time per unit area.

Electricity cost: Cost = rated power of appliance \times duration of use \times cost per unit of

Cost = units of electricity used \times cost per unit of electricity.

Prefixes of the metric system

Factor	Prefix	Symbol
10 ¹⁸	exa	Е
10 ¹⁵	peta	Р
1012	tera	Т
10 ⁹	giga	G
10 ⁶	mega	М
10 ³	kilo	k
10 ⁻³	milli	m
10-6	micro	μ

Standard atomic weights of selected elements

Name	Symbol	Atomic Weight
aluminium	Αℓ	26.98
antimony	Sb	121.76
carbon	С	12.01
copper	Cu	63.55
gold	Au	196.97
hydrogen	Н	1.008
iron	Fe	55.85
lead	Pb	207.2
nickel	Ni	58.69
oxygen	0	16.00
silicon	Si	28.09
sulfur	S	32.07
titanium	Ti	47.87
zinc	Zn	65.41
zirconium	Zr	91.22